This article was downloaded by: [Tomsk State University of Control Systems and Radio]

On: 18 February 2013, At: 11:16

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



# Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gmcl19

On the Mixed Valence Behavior and Cooperative 3D Ordering of a Series of Tris-Oxalato Ferrates:  $Bu_4N\{M Fe^{III}(ox)_3\}$  (M=Mn<sup>II</sup>(A),Fe<sup>II</sup>(B), Co<sup>II</sup>(C), Ni<sup>II</sup>(D) and  $\phi_4P\{Fe^{II}Fe^{III}(ox)_3\}(E)$ : New Ferrimagnets

W. M. Reiff <sup>a</sup> , J. Kreisz <sup>a</sup> , L. Meda <sup>a</sup> & R. U. Kirss <sup>a</sup>

Version of record first published: 24 Sep 2006.

To cite this article: W. M. Reiff , J. Kreisz , L. Meda & R. U. Kirss (1995): On the Mixed Valence Behavior and Cooperative 3D Ordering of a Series of Tris-Oxalato Ferrates:  $Bu_4N\{M Fe^{III}(ox)_3\}$  ( $M=Mn^{II}(A),Fe^{II}(B),Co^{II}(C),Ni^{II}(D)$  and  $\phi_4P\{Fe^{II}Fe^{III}(ox)_3\}(E)$ : New Ferrimagnets, Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 273:1, 181-187

To link to this article: <a href="http://dx.doi.org/10.1080/10587259508031854">http://dx.doi.org/10.1080/10587259508031854</a>

#### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings,

<sup>&</sup>lt;sup>a</sup> Department of Chemistry, Northeastern University, Boston, Mass, 02115

demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Printed in Malaysia

ON THE MIXED VALENCE BEHAVIOR AND COOPERATIVE 3D ORDERING OF A SERIES OF TRIS-OXALATO FERRATES: Bu<sub>4</sub>N{M  $Fe^{II}(ox)_3$ }(M=Mn<sup>II</sup>(A),Fe<sup>II</sup>(B), Co<sup>II</sup>(C), Ni<sup>II</sup>(D) AND  $\phi_4$ P{Fe<sup>II</sup>Fe<sup>III</sup>(ox)<sub>3</sub>}(E): NEW FERRIMAGNETS

W. M. REIFF \*, J. KREISZ, L. MEDA, AND R. U. KIRSS, Department of Chemistry, Northeastern University, Boston, Mass 02115.

Abstract A combination of a.c. susceptometry and Fe<sup>57</sup> Mossbauer spectroscopy measurements confirm valence trapped 3D-ferrimagnetically ordered ground states for the title compounds save (A) that orders antiferromagnetically with a very broad maximum in  $\chi'_m$  centered at  $\sim 50$ K. Their ground state magnetic behavior is contrasted with the previously demonstrated ferromagnetism of Bu<sub>4</sub>N{Fe<sup>II</sup>Cr<sup>III</sup>(ox<sub>3</sub>)} and  $\phi_4$ P{Mn<sup>II</sup>Cr<sup>III</sup>(ox<sub>3</sub>)}.

## INTRODUCTION

As part of our continuing interest in dynamic mixed valence behavior, we have studied the title compounds using Fe<sup>57</sup> Mossbauer spectroscopy and a.c. susceptometry over the range 4.2K to 300K. Unfortunately, we find that all of these materials are valence trapped on the Fe<sup>57</sup> Mossbauer spectroscopy time scale (100ns) showing typical high-spin iron III spectra for (A), (C) and (D) and valence trapped high-spin Fe II and Fe III for (B) and (E). The initial studies (1) in this area focused on the Bu<sub>4</sub>N{MCr(ox)<sub>3</sub>}series, M=Mn<sup>II</sup>, Fe<sup>II</sup>, Co<sup>II</sup>, Ni<sup>II</sup>, and Cu<sup>II</sup> and  $\phi_4$ P{MnCr(ox)<sub>3</sub>}(F), (2), all of which were found to be genuine bulk 3D ferromagnets, albeit, with low T<sub>c</sub>'s ( $\leq$  14K) with the butyl systems assumed to have 3D-network structures.

## **RESULTS**

For the present systems we observe interesting ferrimagnetic behavior for (B), (C) and (D), and  $\phi_4 P\{Fe^{II}Fe^{III}(ox)_3\}(E)$  with high critical temperatures: 44K, 30K, 28K and 37K respectively and antiferro-magnetism for (A). Our powder x-ray diffractometry studies show (E) isomorphous with (F) whose single crystal x-ray study confirms a layered structure. The ferrimagnetism of the present systems is accompanied by strong out of phase absorption  $(\chi_m" \neq 0)$  and is clearly complex for (C) and (E) with multiple transitions in evidence for  $\chi_m'$  and  $\chi_m"$  in the vicinity of  $T_{Critical}$  suggesting possible changes of magnetic structure (spin reorientation) just below the initial ordering temperature. The a.c. susceptibility for the case of (B) is shown in Figure 1.

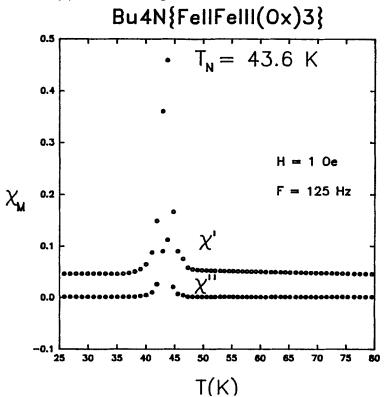


FIGURE 1 Real  $(\chi'_m)$  and imaginary  $\chi''_m$  components of the a.c. susceptibility of Bu<sub>4</sub>N{FeII FeIII(0x)3}

The 4.2K Mossbauer spectrum of  $Bu_4N\{Fe^{II}Cr^{III}(ox)_3\}$  (G) corresponds to the perturbation scheme  $\Delta E(quadrupole) >> H(internal)$  i.e., a low velocity doublet and high velocity triplet indicating that  $V_{zz}$  (the principal component of the electric field gradient tensor) is negative at the ferrous sites (3). This was useful in the deconvolution of the spectra of (B) and (E) whose high spin ferric component exhibits an additional classical six line pattern (i.e., H(internal) >>  $\Delta E(quadrupole)$ , with H(internal)  $\approx 53T$  vs only  $\sim 4T$  for the Fe<sup>II</sup> as seen in Figure 2.

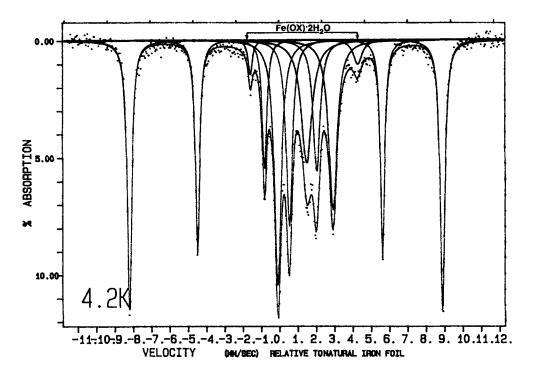


FIGURE 2 Iron-57 Mossbauer spectrum of Bu<sub>4</sub>N {FeII FeIII (ox)<sub>3</sub>} at 4.20K, arrows indicate Fe(ox)·2H<sub>2</sub>O contamination

At low temperatures, Figure 3, a single high spin  $Fe^{III}$  environment is apparent for compounds (A) and (C) with H(internal)  $\approx 52T$ .

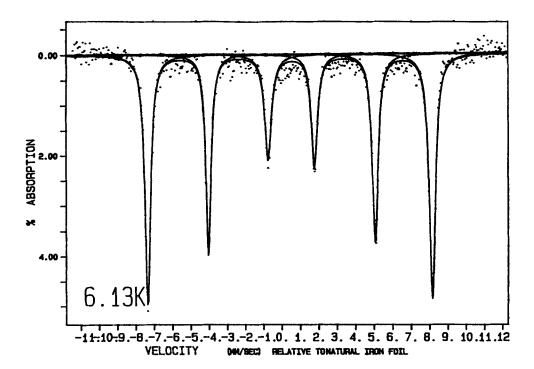


FIGURE 3 Mossbauer spectrum of Bu<sub>4</sub>N{MnIIFeIII(ox)<sub>3</sub>} at 6.13K

The values for the Curie temperatures of (F) and (G) as determined via a.c. susceptometry are 5.6K and 12K in good agreement with those found (5.9K and 12K) using d.c. magnetometry (1, 2). The near zero field a.c. measurements with  $\chi^{"}_{m} \neq 0$  Figures 4 and 5 further confirm the 3D ferromagnetic ground states of these materials and eliminate any possibility of some type of field induced (metamagnetic) behavior in the original studies of these materials, (1) and (2) at 0.1 and 0.2T respectively.

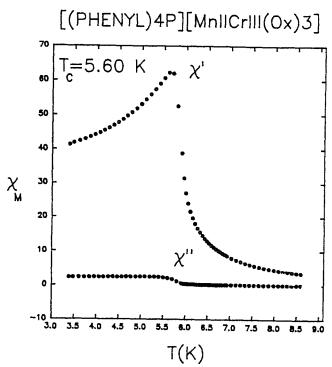


FIGURE 4 The real  $(\chi'_{n})$  and imaginary  $(\chi''_{n})$  components of the a.c. susceptibility of  $\phi_{4}P\{MnIICrIII(ox)_{3}\}$   $Bu4N[Fe||Cr|||(OX)_{3}]$ 

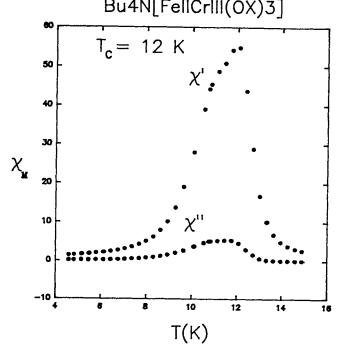


FIGURE 5 Real  $(\chi'_m)$  and imaginary  $(\chi''_m)$  components of the a.c. susceptibility of Bu<sub>4</sub>N{FeII CrIII(0x)<sub>3</sub>}

The shapes of the susceptibility curves (particularly  $\chi_m$ ') for (F) and (G) are quite different. The a.c. response of (G) appears to suggest essentially single domain behavior below low  $T_c$  while the behavior of (F) is more reminiscent of a multidomain material (4). Perhaps this difference reflects a fundamental difference in molecular structures, although the similarity of  $T_c$  for the Bu<sub>4</sub>N<sup>+</sup> and  $\phi_4$ P<sup>+</sup> Mn<sup>II</sup>Cr<sup>III</sup> analogues has been used (2) to suggest a 2D structure for the former.

#### Summary

Finally, we point out that chemical analyses and Mossbauer spectra indicate a small degree of contamination from ( $\sim 2\%$  to 8%) (for compounds (B), (C) and (D)) with the well known chain polymer Fe(ox)  $2H_2O$ . That is, some reduction of the tris-oxalato ferrate III apparently occurs during the preparative stage. This in turn is likely related to photo-sensitivity of  $K_3[M^{III}(ox)_3]$  core systems in general as previously suggested (5) specifically for  $K_3[Co(ox)_3] \cdot 3H_2O$ . In any event, the presence of a few percent Fe(ox)  $2H_2O$  should have no measurable effect on the magnetic properties of the systems reported on herein and in references (1) and (2). Its ground state is 3D antiferromagnetic ( $T_N=11.7K$ ) (6,7). Moreover, it is a highly 1D a.f. chain with strong intra-chain exchange ( $T_x(max) \sim 40K$ ). Thus in the temperature range of relevance (30K to 45K), the present (Co and Ni) ferrimagnets exhibit molar susceptibilities 25 to 50 times greater than that of any Fe(ox)  $2H_2O$  impurity.

## **REFERENCES**

- \* Author to whom correspondence should be addressed, supported by the NSF Division of Materials Research
- 1. H. Tamaki, Z.J. Zhong, N. Matsumoto, S. Kida, M. Koikawa, N. Achiwa, Y. Hashimoto and H. Okawa, J. Am. Chem. Soc., 114, 6974, (1992)
- S. Decurtins, H.W. Schmalle, H.R. Oswald, A. Linden, J. Ensling,
  P. Gütlich and A. Hauser, <u>Inorganica Chimica Acta</u>, 65 216, (1994)
- 3. W.M. Reiff, Coord.Chem.Rev., 2, 10, 37 (1973)
- 4. D.W. Carnegie, Jr., C.J. Tranchita, and H. Claus, J. Appl. Phys., 50 (1979)
- 5. J.C. Bailar and E.M. Jones, <u>Inorganic Syntheses</u>, in H.S. Booth (ed.), (Vol. 1, McGraw-Hill, New York), <u>37</u>, (1939),
- S. Simizu, J.Y. Chen, S.A. Friedberg, J. Martinez and G. Shirane, J. Appl. Phys. 61, (1987)
- 7. J.Y. Chen, S. Simizu and S.A. Friedberg, <u>J. Appl. Phys.</u>, <u>57</u>, (1985)